

Solutions to a) exercises

Please note that just the numerical and short-answer solutions are supplied. For full worked solutions to all of the (a) exercises, please refer to the *Solutions Manual* by Charles Trapp, Marshall Cady, and Carmen Giunta (9780198701286).

Topic 1

E1.1a

	Example	Element	Ground-state electronic configuration
(a)	Group 2	Ca, calcium	[Ar]4s ²
(b)	Group 7	Mn, manganese	[Ar]3d ⁵ 4s ²
(c)	Group 15	As, arsenic	[Ar]3d ¹⁰ 4s ² 4p ³

E1.2a (a) Chemical formula and name: MgCl₂, magnesium chloride. Ions: Mg²⁺ and Cl⁻. Oxidation numbers of the elements: magnesium, +2; chlorine, -1.

(b) Chemical formula and name: FeO, iron(II) oxide. Ions: Fe²⁺ and O²⁻. Oxidation numbers of the elements: iron, +2; oxygen, -2.

(c) Chemical formula and name: Hg₂Cl₂, mercury(I) chloride. Ions: Cl⁻ and Hg₂²⁺ (a polyatomic ion). Oxidation numbers of the elements: mercury, +1; chlorine, -1.

E1.8a

- (a) CO₂ is a linear, nonpolar molecule.
- (b) SO₂ is a bent, polar molecule.
- (c) N₂O is linear, polar molecule.
- (d) SF₄ is a seesaw molecule and it is a polar molecule.

E1.9a In the order of increasing dipole moment: CO₂, N₂O, SF₄, SO₂.

E1.10a

- (a) Mass is an extensive property.
- (b) Mass density is an intensive property.
- (c) Temperature is an intensive property.
- (d) Number density is an intensive property.

E1.11a (a) 0.543 mol (b) 3.27 × 10²³ molecules

E1.12a (a) 108. g (b) 1.77 N

E1.13a 0.43 bar

E1.14a 0.42 atm

E1.15a (a) 1.47×10^5 Pa (b) 1.47 bar

E1.16a $T = 310.2$ K

E1.17a $\theta/^{\circ}\text{C} = \frac{5}{9} \times (\theta_{\text{F}}/^{\circ}\text{F} - 32)$ or $\theta_{\text{F}}/^{\circ}\text{F} = \frac{9}{5} \times \theta/^{\circ}\text{C} + 32$

$$\theta_{\text{F}} = 173 \text{ }^{\circ}\text{F}$$

$$\theta_{\text{R}} = 491.67 \text{ }^{\circ}\text{R}$$

E1.18a 105 kPa

E1.19a S_8

E1.20a 1.8 MPa

E1.21a 4.6×10^5 Pa, 6.9×10^5 Pa

Topic 2

E2.1a (a) 9.81 ms^{-1} , 48 mJ (b) 29.4 ms^{-1} , 0.43 J

E2.2a $s_{\text{terminal}} = \frac{zeE}{6\pi\eta R}$

E2.5a $1.0 \hbar$

E2.6a (a) 2.25×10^{-20} J (b) 9.00×10^{-20} J

E2.7a $1.88 \times 10^8 \text{ m s}^{-1}$, 100 keV

E2.8a 1.15×10^{-18} J, 1.48×10^{-20} J

E2.9a -2.40 V

E2.10a 24.1 kJ, 28.8 $^{\circ}\text{C}$

E2.11a 27.2 K or 27.2 $^{\circ}\text{C}$

E2.12a 128 J

E2.13a $2.4194 \text{ J K}^{-1} \text{ g}^{-1}$

E2.14a $75.3 \text{ J K}^{-1} \text{ mol}^{-1}$

E2.15a $8.3145 \text{ kJ mol}^{-1}$

E2.16a $S_{\text{H}_2\text{O(g)}} > S_{\text{H}_2\text{O(l)}}$

E2.17a $S_{\text{Fe(3000 K)}} > S_{\text{Fe(300 K)}}$

E2.19a (a) 1.6×10^{-17} (b) 0.021

E2.21a 4.631×10^{-6}

E2.23a 1.07

E2.24a 1.25

E2.25a 0.47 kJ

E2.26a (a) 1.38 kJ (b) 4.56 kJ

E2.27a $12.47 \text{ J mol}^{-1} \text{ K}^{-1}$

E2.28a (a) $20.79 \text{ J mol}^{-1} \text{ K}^{-1}$ (b) $24.94 \text{ J mol}^{-1} \text{ K}^{-1}$

Topic 3

E3.1a $2.26 \times 10^8 \text{ m s}^{-1}$

E3.2a $4.00 \text{ }\mu\text{m}$, $7.50 \times 10^{13} \text{ Hz}$

Topic 4

E4.1a (a) 33 zJ, 20 kJ mol^{-1} (b) $3.3 \times 10^{-34} \text{ J}$, 0.20 nJ mol^{-1}

E4.2a (a) No kinetic energy and zero speed (b) 0.452 aJ, 996 km s^{-1}

E4.3a $2.23 \text{ }\mu\text{m}$

E4.4a 6.96 keV without a relativist mass correction ; 6.96 keV with the relativistic mass correction

The relativistic mass correction did not make a difference in this exercise.

E4.6a (a) $3.0 \times 10^{19} \text{ s}^{-1}$ (b) $7.4 \times 10^{20} \text{ s}^{-1}$

E4.7a (a) $3.3 \times 10^{-29} \text{ m}$ (b) $1.3 \times 10^{-36} \text{ m}$ (c) 99.7 pm

E4.8a $7.27 \times 10^6 \text{ m s}^{-1}$, 150 V

E4.9a $1.71 \times 10^6 \text{ m s}^{-1}$

Topic 5

E5.3a $N = \left(\frac{1}{2\pi}\right)^{1/2}$

E5.4a $(1/2\pi) d\phi$

E5.5a $\frac{1}{2}$

Topic 6

E6.1a $\hat{V} = \frac{1}{2}k_f x^2$

E6.2a -4

E6.3a (a) ik (b) not an eigenfunction of d/dx (c) not an eigenfunction of d/dx (d) not an eigenfunction of d/dx (e) not an eigenfunction of d/dx (f) not an eigenfunction of d/dx

Topic 7

E7.3a $\frac{L}{2}$

E7.4a 0

Topic 8

E8.1a 52 pm

E8.2a (a) $1.1 \times 10^{-28} \text{ ms}^{-1}$ (b) $1 \times 10^{-27} \text{ m}$

E8.3a (a) 0 (b) 0 (c) $i\hbar$ (d) $2ix\hbar$ (e) $ni x^{n-1}\hbar$

Topic 9

E9.1a $3 \times 10^{-34} \text{ kg m s}^{-1}$, $5 \times 10^{-38} \text{ J}$

E9.2a Ae^{ikx} , $2.7 \times 10^{33} \text{ m}^{-1}$

E9.3a (a) 1.8×10^{-19} J, 1.1×10^2 kJ mol⁻¹, 1.1 eV, 9.1×10^3 cm⁻¹ (b) 6.6×10^{-19} J, 4.0×10^2 kJ mol⁻¹, 4.1 eV, 3.3×10^4 cm⁻¹

E9.4a (a) 0.04 (b) 0

E9.5a $\frac{h^2}{4L^2}$

E9.6a $\frac{\lambda_c}{8^{1/2}}$

E9.7a $x = \frac{L}{10}, \frac{3L}{10}, \frac{L}{2}, \frac{7L}{10},$ and $\frac{9L}{10}$

Topic 10

E10.1a 0.8

Topic 11

E11.1a (a) 3.6×10^{-19} J, 2.2×10^2 kJ mol⁻¹, 2.3 eV, 1.8×10^4 cm⁻¹ (b) 1.3×10^{-18} J, 8.0×10^2 kJ mol⁻¹, 8.3 eV, 6.7×10^4 cm⁻¹

E11.2a (a) 5.5×10^{14} s⁻¹, 5.5×10^{-7} m, 550 nm (b) 2.0×10^{15} s⁻¹, 1.5×10^{-7} m, 150 nm

E11.3a (a) 5.5×10^3 K (b) $3.\bar{3} \times 10^3$ K

E11.4a (a) 0.0016 (b) 0

E11.5a (1/8, 1/10), (1/8, 3/10), (1/8, 1/2), (1/8, 7/10), (1/8, 9/10)

(3/8, 1/10), (3/8, 3/10), (3/8, 1/2), (3/8, 7/10), (3/8, 9/10)

(5/8, 1/10), (5/8, 3/10), (5/8, 1/2), (5/8, 7/10), (5/8, 9/10)

(7/8, 1/10), (7/8, 3/10), (7/8, 1/2), (7/8, 7/10), (7/8, 9/10)

E11.6a $\frac{x}{L} = 0, \frac{1}{2}, 1, \frac{y}{L} = 0, \frac{1}{3}, \frac{2}{3}, 1$

E11.7a $\frac{h^2}{2L^2}$

E11.8a $\frac{h}{2m_e c}, \frac{\lambda_c}{2}$

E11.10a 3

E11.11a 0.23 or 23 per cent

E11.12a $n = 9.77 \times 10^{10}$, 1.27×10^{-31} J, $m = 25.8$ pm

Topic 12

E12.1a 1.61×10^{-20} J

E12.2a 278 N m^{-1}

E12.3a $2.63 \text{ }\mu\text{m}$

E12.4a $3.72 \text{ }\mu\text{m}$

E12.5a $\pm 0.525\alpha$ or $\pm 1.65\alpha$

E12.7a 565 cm^{-1}

E12.8a 0.056

E12.9a $\langle E_k \rangle = 3\langle V \rangle/2$

Topic 13

n/a

Topic 14

E14.1a 1.49×10^{-34} J s, 0 or $\pm \hbar$

E14.2a (a) $\pi/4$ radians or 45° (b) $\pi/2$ radians or 90°

E14.3a 7

Topic 15

E15.1a 0

E15.2a $\frac{\epsilon}{2}$

E15.3a $\frac{\epsilon}{2}$

E15.4a 0

E15.5a $-\frac{mg^2}{2\omega^2}$

Topic 16E16.1a $4.50 \times 10^{-16} \text{ J m}^{-3} \text{ s}$ **Topic 17**

E17.2a 14.0 eV

E17.3a $N = \frac{2}{a_0^{3/2}}$

E17.4a $r = 4a_0, r = 0$ E17.5a $r = 6a_0 = 3.175 \times 10^{-10} \text{ m}$ **Topic 18**

E18.1a (a) $l = 0$, so angular momentum = 0 (b) $l = 1$, so angular momentum = $\sqrt{2}\hbar$ (c)
 $l = 3$, so angular momentum = $2\sqrt{3}\hbar$

E18.2a 4

E18.3a (a) $g = 1$ (b) $g = 4$ (c) $g = 16$ E18.4a $2E_{1s}, -E_{1s}, -17.4376 \text{ aJ}, 8.7188 \text{ aJ}$

E18.5a $\langle r_{2s} \rangle = \frac{6a_0}{Z}, r^* = (3 + \sqrt{5}) \frac{a}{Z}$

E18.6a $P_{3s} = 4\pi r^2 \left(\frac{1}{4\pi} \right) \times \left(\frac{1}{243} \right) \times \left(\frac{Z}{a_0} \right)^3 \times (6 - 6\rho + \rho^2)^2 e^{-\rho}, r = 0.74a_0/Z, 4.19a_0/Z, \text{ and } 13.08a_0/Z$

E18.7a p_z orbital: The nodal plane is the xy plane and $\theta = \pi/2$ is an angular node. p_x orbital: The nodal plane is the yz plane and $\theta = 0$ is an angular node. p_y orbital: The nodal plane is the xz plane and $\theta = 0$ is an angular node.**Topic 19**

E19.1a $8.89 \times 10^{10} \text{ m s}^{-1}$ **Topic 20**

E20.2a Sc: $[\text{Ar}]4s^23d^1$
 Ti: $[\text{Ar}]4s^23d^2$
 V: $[\text{Ar}]4s^23d^3$
 Cr: $[\text{Ar}]4s^23d^4$ or $[\text{Ar}]4s^13d^5$ (most probable)
 Mn: $[\text{Ar}]4s^23d^5$
 Fe: $[\text{Ar}]4s^23d^6$
 Co: $[\text{Ar}]4s^23d^7$
 Ni: $[\text{Ar}]4s^23d^8$
 Cu: $[\text{Ar}]4s^23d^9$ or $[\text{Ar}]4s^13d^{10}$ (most probable)
 Zn: $[\text{Ar}]4s^23d^{10}$

Topic 21E21.1a $9.118 \times 10^{-6} \text{ cm}$, $1.216 \times 10^{-5} \text{ cm}$ E21.2a $\tilde{\nu} = 3.292 \times 10^5 \text{ cm}^{-1}$, $\lambda = 3.038 \times 10^{-6} \text{ cm}$, $\nu = 9.869 \times 10^{15} \text{ s}^{-1}$

E21.3a (a) forbidden (b) allowed (c) allowed

E21.4a (a) $[\text{Kr}]4d^8$ (b) $S = 1, 0$; $M_S = -1, 0, +1$; $M_S = 0$ E21.5a (a) $\frac{5}{2}, \frac{3}{2}$ (b) $\frac{7}{2}, \frac{5}{2}$ E21.6a $l = 3$ or 2 , $l = 1$ or 0 E21.7a $J = 3, 2, 1$ E21.8a $L = 1, S = 1, J = 2$ E21.9a (a) 1, 0 with multiplicities 3, 1, respectively (b) $S = \frac{3}{2}, \frac{1}{2}$ [from 1], and $\frac{1}{2}$ [from 0], with multiplicities 4, 2, 2E21.10a $^3D_3, ^3D_2, ^3D_1, ^1D_2, ^3D$ set of terms are the lower in energyE21.11a (a) $J = 1$ (3S_1) and there are 3 states (b) $J = \frac{5}{2}, \frac{3}{2}$, respectively ($^2D_{5/2}, ^2D_{3/2}$), with 6, and 2 states, respectively (c) $J = 1$ (1P_1) with 3 statesE21.12a (a) $^2S_{1/2}$ (b) $^2D_{5/2}$ and $^2D_{3/2}$

Topic 22

E22.1a $\{s(1)p_z(2)+s(2)p_z(1)\} \times \{\alpha(1)\beta(2)-\alpha(2)\beta(1)\}$

E22.2a $\psi = a\psi_{\text{VB}} + b\psi_{\text{HF}^+} + c\psi_{\text{HF}^-}$

E22.6a $N = 3^{-1/2}$, $\psi = 3^{-1/2}(s + 2^{1/2}p)$

Topic 23

E23.1a $N = \left(\frac{1}{1+2\lambda S + \lambda^2} \right)^{1/2}$

E23.2a $N = 1.12$, $\psi_1 = 0.163A + 0.947B$, $b = 0.412$, $a = -1.02$, $\psi_2 = -1.02A + 0.412B$

E23.3a 1.9 eV, 130 pm

E23.4a The bonding orbital is odd (u) and the antibonding orbital is even (g).

Topic 24

24.1a (a) $1\sigma_g^2$, $b = 1$ (b) $1\sigma_g^2 1\sigma_u^2$, $b = 0$ (c) $1\sigma_g^2 1\sigma_u^2 1\pi_u^4$, $b = 2$

24.2a C_2

24.3a F_2^+

24.4a $b = 1, b = 0, b = 1, b = 2, b = 3, b = 2, b = 1$

24.5a $2\pi_g$

24.6a $4 \times 10^5 \text{ m s}^{-1}$

24.7a (a) $2.1 \times 10^{-10} \text{ m} = 0.21 \text{ nm}$ (b) $1.0 \times 10^{-10} \text{ m} = 0.10 \text{ nm}$

Topic 25

E25.1a (a) $1\sigma^2 2\sigma^2 1\pi^4 3\sigma^2$ (b) $1\sigma^2 2\sigma^2 3\sigma^2 1\pi^4 2\pi^1$ (c) $1\sigma^2 2\sigma^2 1\pi^4 3\sigma^2$

E25.3a NO^+

E25.5a 7.2, 8.3

E25.6a -6.6 or -8.9

E25.7a -5.0 or -10.7 eV

Topic 26

$$\text{E26.1a (a)} \begin{vmatrix} \alpha - E & \beta & 0 \\ \beta & \alpha - E & \beta \\ 0 & \beta & \alpha - E \end{vmatrix} \quad \text{(b)} \begin{vmatrix} \alpha - E & \beta & \beta \\ \beta & \alpha - E & \beta \\ \beta & \beta & \alpha - E \end{vmatrix}$$

$$\text{E26.2a (a)} \text{C}_6\text{H}_6^- \text{ (7 electrons): } a_{2u}^2 e_{1g}^4 e_{2u}^1 \quad E_\pi = 2(\alpha + 2\beta) + 4(\alpha + \beta) + (\alpha - \beta) = 7\alpha + 7\beta$$

$$\text{(b) C}_6\text{H}_6^+ \text{ (5 electrons): } a_{2u}^2 e_{1g}^3 \quad E_\pi = 2(\alpha + 2\beta) + 3(\alpha + \beta) = 5\alpha + 7\beta$$

$$\text{E26.3a (a)} 7\beta, 0 \quad \text{(b)} 7\beta, 2\beta$$

$$\text{E26.5a (a)} 14\alpha + 19.314\beta \quad \text{(b)} 14\alpha + 19.448\beta$$

Topic 27

$$\text{E27.1a } \hat{V} = -3j_0 \left\{ \frac{1}{r_{Li1}} + \frac{1}{r_{Li2}} + \frac{1}{r_{Li3}} + \frac{1}{r_{Li4}} \right\} - j_0 \left\{ \frac{1}{r_{H1}} + \frac{1}{r_{H2}} + \frac{1}{r_{H3}} + \frac{1}{r_{H4}} \right\} + j_0 \left\{ \frac{1}{r_{12}} + \frac{1}{r_{13}} + \frac{1}{r_{14}} + \frac{1}{r_{23}} + \frac{1}{r_{24}} + \frac{1}{r_{34}} \right\}$$

$$\text{E27.2a } N_e = 2 \text{ for HeH}^+, \quad \hat{H} = -\frac{\hbar^2}{2m_e} \{ \nabla_1^2 + \nabla_2^2 \} - 2j_0 \left\{ \frac{1}{r_{He1}} + \frac{1}{r_{He2}} \right\} - j_0 \left\{ \frac{1}{r_{H1}} + \frac{1}{r_{H2}} \right\} + j_0 \left\{ \frac{1}{r_{12}} \right\}$$

$$\text{E27.3a } N_e = 2 \text{ for HeH}^+, \quad \Psi = \frac{1}{\sqrt{2}} \begin{vmatrix} \psi_a^\alpha(1) & \psi_a^\beta(1) \\ \psi_a^\alpha(2) & \psi_a^\beta(2) \end{vmatrix}$$

$$\text{E27.4a } f_1 \psi_a(1) = h_1 \psi_a(1) + J_a(1) \psi_a(1) = \varepsilon_a \psi_a(1)$$

$$\text{E27.6a } F_{AA} = E_{\text{He}} + c_{\text{Hea}}^2 (\text{HeHe}|\text{HeHe}) + 2c_{\text{Hea}}c_{\text{Ha}} (\text{HeHe}|\text{HeH}) + c_{\text{Ha}}^2 \{ 2(\text{HeHe}|\text{HH}) - (\text{HeH}|\text{HeH}) \}$$

$$F_{AB} = \int \chi_{\text{He}}(1) h_1 \chi_{\text{H}}(1) d\tau_1 + c_{\text{Hea}}^2 (\text{HeH}|\text{HeHe}) + c_{\text{Hea}}c_{\text{Ha}} \{ 3(\text{HeH}|\text{HeH}) - (\text{HeHe}|\text{HH}) \} + c_{\text{Ha}}^2 (\text{HeH}|\text{HH})$$

$$\text{E27.8a (a)} 17 \text{ basis functions} \quad \text{(b)} 28 \text{ basis functions} \quad \text{(c)} 34 \text{ basis functions}$$

Topic 28

$$\text{E28.1a } c_{\text{Ha}} = \left(\frac{1}{1 + \left(\frac{\beta}{\alpha_{\text{He}} - \varepsilon_a} \right)^2} \right)^{1/2}, \quad \varepsilon_a = \frac{\alpha_{\text{He}} + \alpha_{\text{H}} - \left(\{ \alpha_{\text{He}} - \alpha_{\text{H}} \}^2 + 4\beta^2 \right)^{1/2}}{2}, \quad c_{\text{Hea}} = \left(\frac{1}{1 + \left(\frac{\beta}{\alpha_{\text{He}} - \varepsilon_a} \right)^2} \right)^{1/2}$$

Topic 29

$$\begin{aligned}
 \text{E29.2a } \Psi_1(1 \text{ of } 4) &= (1/2)^{1/2} \begin{vmatrix} \psi_b^\alpha(1) & \psi_a^\beta(1) \\ \psi_b^\alpha(2) & \psi_a^\beta(2) \end{vmatrix}, \quad \Psi_1(2 \text{ of } 4) = (1/2)^{1/2} \begin{vmatrix} \psi_a^\alpha(1) & \psi_b^\beta(1) \\ \psi_a^\alpha(2) & \psi_b^\beta(2) \end{vmatrix}, \\
 \Psi_1(3 \text{ of } 4) &= (1/2)^{1/2} \begin{vmatrix} \psi_a^\alpha(1) & \psi_b^\alpha(1) \\ \psi_a^\alpha(2) & \psi_b^\alpha(2) \end{vmatrix}, \quad \Psi_1(4 \text{ of } 4) = (1/2)^{1/2} \begin{vmatrix} \psi_a^\beta(1) & \psi_b^\beta(1) \\ \psi_a^\beta(2) & \psi_b^\beta(2) \end{vmatrix}, \\
 \Psi &= C_0\Psi_0 + C_2\Psi_2 = C_0 \begin{vmatrix} \psi_a^\alpha(1) & \psi_a^\beta(1) \\ \psi_a^\alpha(2) & \psi_a^\beta(2) \end{vmatrix} + C_2 \begin{vmatrix} \psi_b^\alpha(1) & \psi_b^\beta(1) \\ \psi_b^\alpha(2) & \psi_b^\beta(2) \end{vmatrix}
 \end{aligned}$$

E29.3a

$$\Psi = C_0\Psi_0 + C_1\{\Psi_1(1 \text{ of } 2) + \Psi_1(2 \text{ of } 2)\} = C_0 \begin{vmatrix} \psi_a^\alpha(1) & \psi_a^\beta(1) \\ \psi_a^\alpha(2) & \psi_a^\beta(2) \end{vmatrix} + C_1 \left\{ \begin{vmatrix} \psi_b^\alpha(1) & \psi_a^\beta(1) \\ \psi_b^\alpha(2) & \psi_a^\beta(2) \end{vmatrix} + \begin{vmatrix} \psi_a^\alpha(1) & \psi_b^\beta(1) \\ \psi_a^\alpha(2) & \psi_b^\beta(2) \end{vmatrix} \right\}$$

$$\begin{aligned}
 \int \Psi_2 \hat{H}^{(1)} \Psi_0 d\tau &= c_{\text{Hea}}^2 c_{\text{Heb}}^2 (\text{HeHe|HeHe}) + 2(c_{\text{Hea}}^2 c_{\text{Heb}} c_{\text{Hb}} + c_{\text{Hea}} c_{\text{Ha}} c_{\text{Heb}}^2) (\text{HeH|HeHe}) \\
 \text{E29.4a } &+ (c_{\text{Hea}}^2 c_{\text{Hb}}^2 + 2c_{\text{Hea}} c_{\text{Ha}} c_{\text{Heb}} c_{\text{Hb}} + c_{\text{Ha}}^2 c_{\text{Heb}}^2) (\text{HeH|HeH}) + 2c_{\text{Hea}} c_{\text{Ha}} c_{\text{Heb}} c_{\text{Hb}} (\text{HH|HeHe}) \\
 &+ 2(c_{\text{Ha}}^2 c_{\text{Heb}} c_{\text{Hb}} + c_{\text{Hea}} c_{\text{Ha}} c_{\text{Hb}}^2) (\text{HH|HHe}) + c_{\text{Ha}}^2 c_{\text{Hb}}^2 (\text{HH|HH})
 \end{aligned}$$

Topic 30

$$\text{E30.2a } 2\{| \psi_a(\mathbf{r}) |^2 + | \psi_b(\mathbf{r}) |^2 \}$$

Topic 31E31.1a The elements, other than the identity E , are a C_3 axis and three vertical mirror planes σ_v E31.2a There are $3C_2$ axes, a centre of inversion, and $3\sigma_h$ mirror planes

E31.3a (a) Sphere: an infinite number of symmetry axes; therefore R_3 (b) Isosceles triangles: E , C_2 , σ_v , and σ'_v ; therefore C_{2v} (c) D_{3h} (d) $D_{\infty h}$

E31.4a (a) C_{2v} (b) $C_{\infty v}$ (c) C_{3v} (d) D_{2h} E31.5a (a) C_{2v} (b) C_{2h} E31.6a (a) pyridine (C_{2v}) and (b) nitroethane (C_s) are polarE31.8a D_{2h} contains i and C_{3h} contains σ_h **Topic 32**

$$\text{E32.1a } \mathbf{D}(\sigma_h) = \begin{pmatrix} -1 & 0 & 0 & 0 \\ 0 & -1 & 0 & 0 \\ 0 & 0 & -1 & 0 \\ 0 & 0 & 0 & -1 \end{pmatrix}$$

$$\text{E32.2a } \begin{pmatrix} -1 & 0 & 0 & 0 \\ 0 & 0 & -1 & 0 \\ 0 & 0 & 0 & -1 \\ 0 & -1 & 0 & 0 \end{pmatrix} = \mathbf{D}(S_3)$$

E32.4a 3

E32.5a 2

Topic 33

E33.1a Integral must be zero

E33.4a No N orbitals, d_{xy}

E33.6a $2A_1 + B_1 + E$

E33.7a (a) either E_{1u} or A_{2u} (b) B_{3u} (x -polarized), B_{2u} (y -polarized), B_{1u} (z -polarized)

Topic 34

E34.1a ClF_3 , O_3 , H_2O_2

E34.2a 2.8 D

E34.3a -36.1 D, -7.54 D, 36.9 D, 168°

E34.4a 2.19×10^{-36} C m, 0.658 μD

Topic 35

E35.1a 1.07×10^3 kJ mol^{-1}

$$\text{E35.2a } \frac{6l^4 Q_1^2}{\pi \epsilon_0 r^5}$$

E35.3a 289 kJ mol^{-1}

Topic 36

E36.1a 1.24 bar, 1.24 bar

E36.2a 2.75 bar

E36.3a (a) 25 bar (b) 22 bar

E36.4a $7.61 \times 10^{-2} \text{ kg m}^5 \text{ s}^{-2} \text{ mol}^{-2}$, $2.26 \times 10^{-5} \text{ m}^3 \text{ mol}^{-1}$

E36.5a (a) 0.92 (b) $1.6 \text{ dm}^3 \text{ mol}^{-1}$, attractive forces dominate

E36.6a 140 atm

E36.7a $1.41 \times 10^3 \text{ K}$

E36.8a 175 pm

E36.9a $0.046 \text{ dm}^3 \text{ mol}^{-1}$

E36.10a 0.66

Topic 37

E37.1a $N = 4$, 4.01 g cm^{-3}

E37.2a (323) and (110)

E37.3a 229 pm, 397 pm, 115 pm

E37.4a 220 pm

E37.5a 291 pm, 10.1° , 14.4° , 17.7°

E37.6a 8.16° , 4.82° , 11.75°

E37.7a 2.14°

E37.8a $f_{\text{Br}^-} = 36$

E37.9a f

E37.10a $3f$ for $h+k$ even and $-f$ for $h+k$ odd

E37.14a 6.1 km s^{-1}

E37.15a 233 pm

Topic 38

E38.1a 0.9069

E38.2a (a) 0.5236 (b) 0.6802 (c) 0.7405

E38.3a (a) 74.9 pm (b) 132 pm

E38.4a Expansion

E38.5a 3500 kJ mol⁻¹

Topic 39

E39.1a 3.54 eV

E39.2a Three unpaired spins

E39.3a $-6.4 \times 10^{-5} \text{ cm}^3 \text{ mol}^{-1}$, $-6.4 \times 10^{-11} \text{ m}^3 \text{ mol}^{-1}$

E39.4a 4.326, 5

E39.5a $+1.6 \times 10^{-8} \text{ m}^3 \text{ mol}^{-1}$

Topic 40

E40.1a 82.9 per cent

E40.2a $5.34 \times 10^3 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$

E40.3a $1.09 \text{ mmol dm}^{-3}$

E40.4a $1.3 \times 10^8 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-2}$

E40.5a $450 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-2}$

E40.6a $159 \text{ dm}^3 \text{ mol}^{-1} \text{ cm}^{-1}$, 23 per cent

E40.7a (a) 0.87 m (b) 2.9 m

E40.8a $0.9999 \times \lambda$

E40.9a (a) 27 ps (b) 2.7 ps

E40.10a (a) $\delta \tilde{\nu} = 53 \text{ cm}^{-1}$ (b) $\delta \tilde{\nu} = 0.53 \text{ cm}^{-1}$

Topic 41

E41.1a $6.33 \times 10^{-46} \text{ kg m}^2$, 0.4421 cm^{-1}

E41.3a (a) asymmetric (b) oblate symmetric (c) spherical (d) prolate symmetric

E41.4a 106.5 pm, 115.6 pm

Topic 42

E42.2a $3.07 \times 10^{11} \text{ Hz}$

E42.3a 127.4 pm

E42.4a $4.442 \times 10^{-47} \text{ kg m}^2$, 165.9 pm

E42.5a (c) CH_4 is inactive.

E42.6a $20\,475 \text{ cm}^{-1}$

E42.7a 198.9 pm

E42.8a $\frac{5}{3}$

Topic 43

E43.1a 16 N m^{-1}

E43.2a 1.077 percent

E43.3a 328.7 N m^{-1}

E43.4a 967.0, 515.6, 411.8, 314.2

E43.5a 1580.38 cm^{-1} , 7.644×10^{-3}

Topic 44

E44.1a (b) HCl (c) CO_2 (d) H_2O

E44.2a (a) 3 (b) 6 (c) 12

E44.3a 127

E44.4a $\frac{1}{2}(\tilde{\nu}_1 + \tilde{\nu}_2 + \tilde{\nu}_3)$

E44.5a (a) Nonlinear: all modes are both infrared and Raman active. (b) Linear: the symmetric stretch is infrared inactive but Raman active.

E44.6a Raman active

E44.7a $4A_1 + A_2 + 2B_1 + 2B_2$

E44.8a A_1, B_1 and B_2 are infrared active; all modes are Raman active.

Topic 45

E45.1a 3, u

E45.2a (a) allowed (b) allowed (c) forbidden (d) forbidden (e) allowed

$$E45.3a \frac{2\sqrt{2}}{3} e^{-2a\tilde{\nu}_0^2/3}$$

$$E45.4a \frac{1}{32} \left(3 + \frac{4}{\pi} \right)^2$$

$$E45.5a \frac{1}{2} (\tilde{B}' + \tilde{B}) / (\tilde{B}' - \tilde{B})$$

$$E45.6a \frac{1}{2} (\tilde{B}' + \tilde{B}) / |(\tilde{B}' - \tilde{B})| - 1, 7$$

E45.7a $30.4 \text{ cm}^{-1} < \tilde{B}' < 40.5 \text{ cm}^{-1}$, greater

E45.8a $\Delta_0 = P - \tilde{\nu}$, $14 \times 10^3 \text{ cm}^{-1}$

$$E45.9a \frac{3}{8} \left(\frac{a^3}{b - \frac{1}{2}a} \right)^{1/2}$$

$$E45.10a \frac{1}{4} e^{-1/16} a$$

Topic 46

E46.1a (a) lower, $\tilde{\nu} \approx 1800 \text{ cm}^{-1}$ (b) no information

E46.3a 20 ps, 70 MHz

Topic 47

E47.1a $\text{s}^{-1} \text{ T}^{-1}$

E47.2a $9.133 \times 10^{-35} \text{ J s}$, $\pm 0.9553 \text{ rad} = \pm 54.74^\circ$

E47.3a 574 MHz

E47.4a $-1.473 \times 10^{-26} \text{ J} \times m_l$

E47.5a 165 MHz

E47.6a (a) $3.98 \times 10^{-25} \text{ J}$ (b) $6.11 \times 10^{-26} \text{ J}$, thus the separation in energy is larger for the proton (a).

E47.7a 16.25 T

E47.9a (a) 1×10^{-6} (b) 5.1×10^{-6} (c) 3.4×10^{-5}

E47.10a 13

E47.11a (a) $2 \times 10^2 \text{ T}$ (b) 10 mT

Topic 48

E48.1a (a) independent (b) 13

E48.2a (a) 11 μT (b) 110 μT

E48.5a 753 MHz

E48.7a 0.39 ms, $2.6 \times 10^3 \text{ s}^{-1}$

Topic 49

E49.1a $5.9 \times 10^{-4} \text{ T}$, 20 μs

Topic 50

E50.1a 2.0022

E50.2a 2.3 mT, $2.002\bar{5}$

E50.3a equal intensity, 330.2 mT, 332.2 mT, 332.8 mT, 334.8 mT

E50.5a (a) 332.3 mT (b) 1.206 T

E50.6a $I = \frac{3}{2}$

Topic 51

E51.1a 21 621 600

E51.2a (a) 40 320 (b) 5.63×10^3 (c) 3.99×10^4

E51.3a 1

E51.4a 524 K

E51.5a 7.43

E51.6a $35\bar{4}$ K

Topic 52

E52.1a (a) (i) 8.23 pm (ii) 2.60 pm (b) (i) 1.79×10^{27} (ii) 5.67×10^{28}

E52.2a 0.3574

E52.3a 72.2

E52.4a (a) 7.97×10^3 (b) 1.12×10^4

E52.5a 18 K

E52.6a 37 K

E52.7a 4.5 K

E52.8a (a) 1 (b) 2 (c) 2 (d) 12 (e) 3

E52.9a 660.6

E52.10a 4500 K

E52.11a 2.571

E52.12a 42.3

E52.13a 4.292, 0.0353 to 0.0377 to 1

Topic 53

E53.1a 8.16×10^{-22} J

E53.2a 18.5 K

E53.3a 25 K

E53.4a 4.5 K

E53.5a 4600 K

E53.6a 10 500 K

E53.7a 6500 K

E53.8a 4.033×10^{-21} J

Topic 54

E54.1a (a) He gas (b) CO gas (c) H₂O vapour

Topic 55

E55.1a (a) $\frac{7}{2}R$, 8.671 kJ mol⁻¹ (b) $3R$, 7.436 kJ mol⁻¹ (c) $7R$, 17.35 kJ mol⁻¹

E55.2a (a) pressure, (b) temperature, and (d) enthalpy are state functions.

E55.3a 3.5×10^3 J needed, 5.7×10^2 J needed

E55.4a -75 J

E55.5a (a) -2.68 kJ, +2.68 kJ (b) -1.62 kJ, +1.62 kJ (c) $w = 0$, 0

E55.6a 1.33 atm, +1.25 kJ, $w = 0$, +1.25 kJ

Topic 56

E56.1a $30 \text{ J K}^{-1} \text{ mol}^{-1}$, $22 \text{ J K}^{-1} \text{ mol}^{-1}$

E56.2a (a) +10.7 kJ, -0.624 kJ, +10.1 kJ (b) +10.7 kJ, +10.1 kJ, +10.1 kJ

E56.3a (a) +2.2 kJ, +2.2 kJ, +1.6 kJ

E56.4a $-7.2 \text{ J atm}^{-1} \text{ mol}^{-1}$, +6.1 kJ

Topic 57

E57.1a 22.5 kJ, -1.6 kJ, 20.9 kJ

E57.2a $-4564.7 \text{ kJ mol}^{-1}$

E57.3a $+53 \text{ kJ mol}^{-1}$, -33 kJ mol^{-1}

E57.4a -167 kJ mol^{-1}

E57.5a $-5152 \text{ kJ mol}^{-1}$, 1.58 kJ K^{-1} , $+3.08 \text{ K}$

E57.6a (a) $-114.40 \text{ kJ mol}^{-1}$, $-111.92 \text{ kJ mol}^{-1}$ (b) $-92.31 \text{ kJ mol}^{-1}$, $-241.82 \text{ kJ mol}^{-1}$

E57.7a $-1368 \text{ kJ mol}^{-1}$

E57.8a (a) $+131.29 \text{ kJ mol}^{-1}$, $+128.81 \text{ kJ mol}^{-1}$ (b) $+134.14 \text{ kJ mol}^{-1}$, $+130.17 \text{ kJ mol}^{-1}$

E57.9a $-803.07 \text{ kJ mol}^{-1}$

E57.10a $-1892 \text{ kJ mol}^{-1}$

Topic 58

E58.1a (a) 0.236 (b) 0.193

E58.2a

	$C_{V,m}$	$C_{p,m}$	γ	Exptl	
$\text{NH}_3(v_v^*=0)$	$3R$	$4R$	1.33	1.31	closer
$\text{NH}_3(v_v^*=6)$	$9R$	$10R$	1.11		
$\text{CH}_4(v_v^*=0)$	$3R$	$4R$	1.33	1.31	closer
$\text{CH}_4(v_v^*=9)$	$12R$	$13R$	1.08		

E58.3a $2.91 \times 10^{-21} \text{ J}$

E58.4a $13\bar{1} \text{ K}$

E58.5a $0.0084\bar{6} \text{ m}^3$, $25\bar{7} \text{ K}$, $-0.89 \times 10^3 \text{ J}$

E58.6a -194 J

E58.7a 9.7 kPa

E58.8a 5.03 mbar

E58.9a $+130.\bar{1} \text{ J mol}^{-1}$, $+7.52 \times 10^3 \text{ J mol}^{-1}$, $-7.39 \times 10^3 \text{ J mol}^{-1}$

E58.10a $1.31 \times 10^{-3} \text{ K}^{-1}$

E58.11a $2.0 \times 10^3 \text{ atm}$

Topic 59

E59.1a Not spontaneous

Topic 60

E60.1a (a) $126 \text{ J K}^{-1} \text{ mol}^{-1}$ (b) $169 \text{ J K}^{-1} \text{ mol}^{-1}$

E60.2a $T = 2.35 \times 10^3 \text{ K}$

E60.3a 43.1, 22.36 K, $43.76 \text{ J K}^{-1} \text{ mol}^{-1}$

E60.4a $54.72 \text{ J K}^{-1} \text{ mol}^{-1}$

E60.5a $11.5 \text{ J K}^{-1} \text{ mol}^{-1}$

E60.6a (a) $34.72 \text{ J K}^{-1} \text{ mol}^{-1}$ (b) $119.06 \text{ J K}^{-1} \text{ mol}^{-1}$

Topic 61

E61.1a $T_c = 191.2 \text{ K}$

E61.2a (a) 366 J K^{-1} (b) 309 J K^{-1}

Topic 62

E62.1a $\text{I}_2(\text{g})$

E62.2a 3.1 J K^{-1}

E62.3a 30.0 kJ mol^{-1}

E62.4a $152.67 \text{ J K}^{-1} \text{ mol}^{-1}$

E62.6a 0, 0

E62.7a $\Delta H = 0, +2.7 \text{ J K}^{-1}$

E62.8a (a) $+2.9 \text{ J K}^{-1}, -2.9 \text{ J K}^{-1}, 0$ (b) $+2.9 \text{ J K}^{-1}, 0, +2.9 \text{ J K}^{-1}$ (c) 0, 0, 0

E62.9a (a) $+87.8 \text{ J K}^{-1} \text{ mol}^{-1}$ (b) $-87.8 \text{ J K}^{-1} \text{ mol}^{-1}$

E62.10a $\Delta S = 92.2 \text{ J K}^{-1}$

Topic 63

E63.1a (a) $9.13 \text{ J K}^{-1} \text{ mol}^{-1}$, $13.4 \text{ J K}^{-1} \text{ mol}^{-1}$, $14.9 \text{ J K}^{-1} \text{ mol}^{-1}$

E63.2a (a) $-386.1 \text{ J K}^{-1} \text{ mol}^{-1}$, (b) $+92.6 \text{ J K}^{-1} \text{ mol}^{-1}$ (c) $-153.1 \text{ J K}^{-1} \text{ mol}^{-1}$

Topic 64

E64.1a (a) $-521.5 \text{ kJ mol}^{-1}$ (b) $+25.8 \text{ kJ mol}^{-1}$ (c) $-178.7 \text{ kJ mol}^{-1}$

E64.2a $-480.98 \text{ kJ mol}^{-1}$

E64.3a $817.90 \text{ kJ mol}^{-1}$

E64.4a $-13.8 \text{ kJ mol}^{-1}$, $-0.20 \text{ kJ mol}^{-1}$

E64.5a (a) $-6.42 \text{ kJ mol}^{-1}$ (b) $-14.0 \text{ kJ mol}^{-1}$

Topic 65

E65.1a (a) $-522.1 \text{ kJ mol}^{-1}$ (b) $+25.78 \text{ kJ mol}^{-1}$ (c) $-178.6 \text{ kJ mol}^{-1}$

E65.2a -340 kJ mol^{-1}

Topic 66

E66.1a -17 J

E66.2a -36.5 J K^{-1}

E66.3a $+10 \text{ kJ}$, 1.6 kJ mol^{-1}

E66.4a $+11 \text{ kJ mol}^{-1}$

E66.5a 12 kJ

Topic 67

E67.1a $P = 1$, $P = 3$

Topic 69

E69.1a $\mu_{\text{W}}(\text{s}) = \mu_{\text{W}}(\text{l}), \mu_{\text{E}}(\text{s}) = \mu_{\text{E}}(\text{l})$

E69.2a -0.38 J mol^{-1}

E69.3a $-3 \times 10^2 \text{ J mol}^{-1}$

E69.4a $1.79 \times 10^5 \text{ J mol}^{-1}$

E69.5a $+45 \text{ J K}^{-1} \text{ mol}^{-1}, +1.59 \times 10^4 \text{ J mol}^{-1} = +15.9 \text{ kJ mol}^{-1}$

E69.6a $+12487 \text{ J mol}^{-1} = +12.487 \text{ kJ mol}^{-1}$

E69.7a 37°C

E69.8a (a) $2.97 \times 10^4 \text{ J mol}^{-1} = 29.7 \text{ kJ mol}^{-1}$ (b) 0.169 bar, 0.82 bar

E69.9a $-0.35^\circ\text{C}, 272.80 \text{ K}$

Topic 70

E70.1a (a) $1.1 \times 10^3 \text{ g} = 1.1 \text{ kg}$ (b) $2.0 \times 10^4 \text{ g}, 20. \text{ kg}$

E70.2a $-1.04 \text{ kJ}, +3.5 \text{ J K}^{-1}$

E70.3a $-1.37 \times 10^3 \text{ J mol}^{-1} = -1.37 \text{ kJ mol}^{-1}$

E70.4a $\frac{1}{2}, 0.4624, 0.5376$

E70.5a $6.4 \times 10^3 \text{ kPa}$

E70.6a 0.268, 0.732, 58.6 kPa

E70.7a $2.0 \times 10^2 \text{ kPa}$

E70.8a 85 g mol^{-1}

E70.9a (a) $3.4 \times 10^{-3} \text{ mol kg}^{-1}$ (b) $3.4 \times 10^{-2} \text{ mol kg}^{-1}$

Topic 71

E71.1a 88.2 kg mol^{-1}

E71.2a (a) 18 kg mol^{-1} (b) 20 kg mol^{-1} (c) 1.1

Topic 72

E72.1a 2.8, phase separation does occur

 E72.2a 0.833, $0.9\bar{3}$, 0.125, $1.2\bar{5}$, $2.\bar{8}$, $1.2\bar{5}$

E72.3a 0.498, 0.667, 1.24, 1.11

E72.4a 0.694

E72.5a 1.55

E72.6a (a) 8.75 g (b) 9.35 g

E72.7a 0.0190, 0.72, 0.85

E72.8a 2.01

Topic 73

$$\text{E73.1a (a) (i) } K = \frac{a_{\text{COCl(g)}} a_{\text{Cl(g)}}}{a_{\text{CO(g)}} a_{\text{Cl}_2(\text{g})}} \quad \text{(ii) } K = \frac{p_{\text{COCl}} p_{\text{Cl}}}{p_{\text{CO}} p_{\text{Cl}_2}}$$

$$\text{(b) (i) } K = \frac{a_{\text{SO}_3(\text{g})}^2}{a_{\text{SO}_2(\text{g})}^2 a_{\text{O}_2(\text{g})}} \quad \text{(ii) } K = \frac{p_{\text{SO}_3}^2 p^{\ominus}}{p_{\text{SO}_2}^2 p_{\text{O}_2}}$$

$$\text{(c) (i) } K = \frac{a_{\text{FeSO}_4(\text{aq})}}{a_{\text{PbSO}_4(\text{aq})}} \quad \text{(ii) } K = \frac{\gamma_{\text{FeSO}_4} b_{\text{FeSO}_4}}{\gamma_{\text{PbSO}_4} b_{\text{PbSO}_4}}$$

$$\text{(d) (i) } K = \frac{a_{\text{HCl(aq)}}^2}{a_{\text{H}_2(\text{g})}} \quad \text{(ii) } K = \frac{\gamma_{\text{HCl(aq)}}^2 b_{\text{HCl(aq)}}^2}{p_{\text{H}_2(\text{g})}} \times \frac{p^{\ominus}}{(b^{\ominus})^2}$$

$$\text{(e) (i) } K = \frac{a_{\text{CuCl}_2(\text{aq})}}{a_{\text{CuCl(aq)}}^2} \quad \text{(ii) } K = \frac{\gamma_{\text{CuCl}_2} b_{\text{CuCl}_2} b^{\ominus}}{\gamma_{\text{CuCl}}^2 b_{\text{CuCl}}^2}$$

 E73.2a $v_{\text{Hg}_2\text{Cl}_2} = -1$, $v_{\text{H}_2} = -1$, $v_{\text{HCl}} = 2$, $v_{\text{Hg}} = 2$

 E73.3a $+12.3 \text{ kJ mol}^{-1}$

 E73.4a $-11.20 \text{ kJ mol}^{-1}$

 E73.5a $-1108 \text{ kJ mol}^{-1}$

 E73.6a (a) 0.087, 0.370, 0.196, 0.348, 1.001 (b) 0.33 (c) 0.33 (d) $+2.8 \text{ kJ mol}^{-1}$

 E73.7a -30 kJ mol^{-1}

Topic 74

74.1a 0.003 725

Topic 75E75.1a K_x is reduced by 67 per centE75.2a 0.045, 1500 K E75.3a $+3.47 \text{ kJ mol}^{-1}$, $-14.8 \text{ J K}^{-1} \text{ mol}^{-1}$ E75.4a (a) $+53 \text{ kJ mol}^{-1}$ (b) -53 kJ mol^{-1} E75.5a 1110 K E75.6a (a) $-68.26 \text{ kJ mol}^{-1}$, 9.2×10^{11} (b) $-63.99 \text{ kJ mol}^{-1}$, 1.3×10^9 , $-69.7 \text{ kJ mol}^{-1}$ **Topic 76**E76.1a (a) $+0.80 \text{ V}$, -0.76 V , $+1.56 \text{ V}$ (b) 0 , -0.40 V , $+0.40 \text{ V}$ (c) -0.74 V , $+0.36 \text{ V}$, -1.10 V (d) $+0.15 \text{ V}$, $+0.77 \text{ V}$, -0.62 V **Topic 77**E77.1a -1.46 V

E77.2a Elemental mercury cannot spontaneously displace the zinc(II) cation from solution under standard conditions.

E77.3a (a) $9.2 \times 10^{-9} \text{ M}$ (b) 8.3×10^{-17} E77.4a -52 kJ mol^{-1} , $-58 \text{ J K}^{-1} \text{ mol}^{-1}$, -69 kJ mol^{-1}

E77.5a

	ν	E_R^\ominus / V	E_L^\ominus / V	$E_{\text{cell}}^\ominus / \text{V}$	$\Delta_r G_{\text{cell}}^\ominus / (\text{kJ mol}^{-1})$
(a)	2	+0.80	-0.76	+1.56	-301
(b)	2	0.00	-0.40	+0.40	-77
(c)	3	-0.74	+0.36	-1.10	+318
(d)	2	+0.15	+0.77	-0.62	+120

E77.6a (a) -0.13 V , -0.44 V , $+0.31\text{ V}$ (b) $+0.27\text{ V}$, 0 , $+0.27\text{ V}$ (c) $+1.23\text{ V}$, 0 , $+1.23\text{ V}$

E77.7a (a) $\text{Cd}^{2+}(\text{aq}) + 2\text{Br}^{-}(\text{aq}) + 2\text{Ag}(\text{s}) \rightarrow \text{Cd}(\text{s}) + 2\text{AgBr}(\text{s})$ (c) -0.62 V

E77.8a (a) 6.4×10^9 (b) 8.9×10^{43}

Topic 78

E78.1a (a) 9.975 (b) 1

E78.2a 1904 m s^{-1} , 478 m s^{-1}

E78.3a 0.69 per cent

E78.4a 333 m s^{-1} , 375 m s^{-1} , 530 m s^{-1}

E78.5a (a) 475 m s^{-1} (b) 82.9 nm (c) $8.10 \times 10^9\text{ s}^{-1}$

E78.6a 0.195 Pa

E78.7a $1.38 \times 10^{-6}\text{ m}$

E78.8a (a) $5.3 \times 10^{10}\text{ s}^{-1}$ (b) $5.3 \times 10^9\text{ s}^{-1}$ (c) $5.3 \times 10^3\text{ s}^{-1}$

E78.9a 1.2×10^{21}

E78.10a 537 s

E78.11a 16 mg

E78.12a 319 g mol^{-1}

E78.13a 1.3 days

Topic 79

E79.1a $7.6 \times 10^{-3}\text{ J K}^{-1}\text{ m}^{-1}\text{ s}^{-1}$

E79.2a (a) $D = 1.5\text{ m}^2\text{ s}^{-1}$, $J_z/N_A = -61\text{ mol m}^{-2}\text{ s}^{-1}$ (b) $D = 1.5 \times 10^{-5}\text{ m}^2\text{ s}^{-1}$, $J_z/N_A = -6.1 \times 10^{-4}\text{ mol m}^{-2}\text{ s}^{-1}$
 (c) $D = 1.5 \times 10^{-7}\text{ m}^2\text{ s}^{-1}$, $J_z/N_A = -6.1 \times 10^{-6}\text{ mol m}^{-2}\text{ s}^{-1}$

E79.3a $-0.078\text{ J m}^{-2}\text{ s}^{-1}$

E79.4a 0.0795 nm^2

E79.5a 103 W

E79.6a 0.201 nm^2

E79.7a (a) $\eta = 178 \mu\text{P}$ (b) $\eta = 186 \mu\text{P}$ (c) $\eta = 342 \mu\text{P}$

Topic 80

E80.1a 16.8 J mol^{-1}

E80.2a $7.63 \times 10^{-3} \text{ S m}^2 \text{ mol}^{-1}$

E80.3a $283 \mu\text{m s}^{-1}$

E80.4a $13.87 \text{ mS m}^2 \text{ mol}^{-1}$

E80.5a $4.01 \times 10^{-8} \text{ m}^2 \text{ V}^{-1} \text{ s}^{-1}$, $5.19 \times 10^{-8} \text{ m}^2 \text{ V}^{-1} \text{ s}^{-1}$, $7.62 \times 10^{-8} \text{ m}^2 \text{ V}^{-1} \text{ s}^{-1}$

E80.6a 420 pm

Topic 81

81.1a $6.2 \times 10^3 \text{ s}$

81.2a (a) 0.00 mol dm^{-3} (b) $0.0121 \text{ mol dm}^{-3}$

Topic 82

E82.1a No change in pressure

E82.2a $8.1 \text{ mol dm}^{-3} \text{ s}^{-1}$, $2.7 \text{ mol dm}^{-3} \text{ s}^{-1}$, $2.7 \text{ mol dm}^{-3} \text{ s}^{-1}$, $5.4 \text{ mol dm}^{-3} \text{ s}^{-1}$

E82.3a $1.35 \overline{5} \text{ mol dm}^{-3} \text{ s}^{-1}$, $4.05 \overline{5} \text{ mol dm}^{-3} \text{ s}^{-1}$, $2.7 \text{ mol dm}^{-3} \text{ s}^{-1}$, $1.35 \overline{5} \text{ mol dm}^{-3} \text{ s}^{-1}$

E82.4a $\text{dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$ (a) $k_r[\text{A}][\text{B}]$ (b) $3k_r[\text{A}][\text{B}]$

E82.5a $\frac{1}{2}k_r[\text{A}][\text{B}][\text{C}]$, $\text{dm}^6 \text{ mol}^{-2} \text{ s}^{-1}$

E82.6a (a) $[k_r] = \text{dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$, $[k_r] = \text{dm}^6 \text{ mol}^{-2} \text{ s}^{-1}$ (b) $[k_r] = \text{kPa}^{-1} \text{ s}^{-1}$, $[k_r] = \text{kPa}^{-2} \text{ s}^{-1}$

Topic 83

E83.1a $n = 2$

E83.2a $1.03 \times 10^4 \text{ s}$ (a) 498 Torr (b) 461 Torr

E83.3a (a) $0.098 \text{ mol dm}^{-3}$ (b) $0.050 \text{ mol dm}^{-3}$

E83.4a $1.11 \times 10^5 \text{ s} = 1.28 \text{ days}$

Topic 84

E84.1a $4.0 \times 10^{10} \text{ dm}^3 \text{ mol s}^{-1}$, $7.1 \times 10^5 \text{ s}^{-1}$, $1.28 \times 10^4 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$

Topic 85

E85.1a $1.08 \times 10^5 \text{ J mol}^{-1} = 108 \text{ kJ mol}^{-1}$, $6.50 \times 10^{15} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$, $6.50 \times 10^{15} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$

E85.2a $3.49 \times 10^4 \text{ J mol}^{-1} = 34.9 \text{ kJ mol}^{-1}$

Topic 86

E86.1a (i) $k_2 K^{1/2} [\text{A}_2]^{1/2} [\text{B}]$ (ii) $\frac{k_2^2 [\text{B}]^2}{4k_1'} \left\{ \left(1 + \frac{16k_1' k_1 [\text{A}_2]}{k_2^2 [\text{B}]^2} \right)^{1/2} - 1 \right\}$, $k_2 K^{1/2} [\text{A}_2]^{1/2} [\text{B}]$, $2k_1 [\text{A}_2]$

E86.2a -3 kJ mol^{-1}

Topic 87

E87.1a $1.13 \times 10^{10} \text{ s}^{-1}$, $1.62 \times 10^{35} \text{ s}^{-1} \text{ m}^{-3}$, 1.7 per cent

E87.2a (a) (i) 1.04×10^{-3} (ii) 0.069 (b) (i) 1.19×10^{-15} (ii) 1.57×10^{-6}

E87.3a (a) (i) 22 per cent (ii) 3 per cent (b) (i) 170 per cent (ii) 16 per cent

E87.4a $1.03 \times 10^{-5} \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1} = 1.03 \times 10^{-2} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$

E87.5a 0.79 nm^2 , 1.16×10^{-3}

Topic 88

E88.1a $4.5 \times 10^7 \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1} = 4.5 \times 10^{10} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$

E88.2a (a) $6.61 \times 10^6 \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1} = 6.61 \times 10^9 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$ (b) $3.0 \times 10^7 \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1} = 3.0 \times 10^{10} \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$

E88.3a $8.0 \times 10^6 \text{ m}^3 \text{ mol}^{-1} \text{ s}^{-1} = 8.0 \times 10^9 \text{ dm}^3 \text{ mol}^{-1} \text{ s}^{-1}$, $4.2 \times 10^{-8} \text{ s}$

E88.4a $1.81 \times 10^8 \text{ mol dm}^{-3} \text{ s}^{-1}$

Topic 89

E89.1a $+69.7 \text{ kJ mol}^{-1}$, $-25 \text{ J K}^{-1} \text{ mol}^{-1}$

E89.2a $+73.4 \text{ kJ mol}^{-1}$, $+71.9 \text{ kJ mol}^{-1}$

E89.3a $-91 \text{ J K}^{-1} \text{ mol}^{-1}$

E89.4a $-74 \text{ J K}^{-1} \text{ mol}^{-1}$

E89.5a (a) $-46 \text{ J K}^{-1} \text{ mol}^{-1}$ (b) $+5.0 \text{ kJ mol}^{-1}$ (c) $+18.7 \text{ kJ mol}^{-1}$

E89.6a $7.1 \text{ dm}^6 \text{ mol}^{-2} \text{ min}^{-1}$

Topic 90

E90.1a Reactant is high in translational energy and low in vibrational energy. Product is high in vibrational energy and relatively lower in translational energy.

Topic 91

E91.1a $1.9 \times 10^{-6} \text{ Pa}^{-1} \text{ s}^{-1}$ or $1.9 \text{ MPa}^{-1} \text{ s}^{-1}$

E91.2a 0.73

E91.3a 5.1×10^{-7}

Topic 92

E92.1a $\frac{k_b K [\text{AH}]^2 [\text{B}]}{[\text{BH}^+]}$

E92.2a $1.50 \text{ mmol dm}^{-3} \text{ s}^{-1}$

E92.3a $2.0 \times 10^{-5} \text{ mol dm}^{-3}$

Topic 93

E93.1a 3.3×10^{18}

E93.2a 0.56 mol dm^{-3}

Topic 94

E94.1a $4 \times 10^{-21} \text{ J}$ or 2 kJ mol^{-1}

E94.2a 12.5 nm^{-1}

Topic 95

E95.1a (i) $1.4 \times 10^{14} \text{ cm}^{-2} \text{ s}^{-1}$ (ii) $3.1 \times 10^{13} \text{ cm}^{-2} \text{ s}^{-1}$

E95.2a 0.13 bar

Topic 96

E96.1a 33.6 cm^3

E96.2a Chemisorption, 50 s

E96.3a $0.83, 0.36$

E96.4a (a) 0.24 kPa (b) 25 kPa

E96.5a 15 kPa

E96.6a $-12.4 \text{ kJ mol}^{-1}$

E96.7a 65 kJ mol^{-1} (a) $1.6 \times 10^{97} \text{ min}$ (b) $2.8 \times 10^{-6} \text{ min}$

E96.8a 61 kJ mol^{-1}

E96.9a (a) $9.1 \text{ ps}, 0.60 \text{ ps}$ (b) $4.1 \times 10^6 \text{ s}, 6.6 \text{ }\mu\text{s}$

E96.10a (a) zeroth-order (b) first-order

Topic 97

E97.1a 12 m^2